

RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College under University of Calcutta)

SECOND YEAR

B.A./B.SC. THIRD SEMESTER (July – December) 2014

Mid-Semester Examination, September 2014

INDUSTRIAL CHEMISTRY (Honours)

Date : 15/09/2014

Time : 2 pm – 4 pm

Paper : III

Full Marks : 50

[Use a separate answer book for each group]

Group – A

1. Define/explain the following terms (**any five**) : [5×1]
 - a) What is china clay?
 - b) What is chemical compositions of clay?
 - c) What is 'glass'?
 - d) What is the function of a 'glaze'?
 - e) How do 'earthenware' differ from 'stoneware'?
 - f) What is refractoriness?
 - g) What is a softening temperature?

2. Answer **any four** : [4×5]
 - a) What is meant by plasticity of clays? Explain the terms—
 - i) Plastic limit of clay
 - ii) Plasticity index and liquid limit. Explain the effect of plasticity of clay on moulding. What is a secondary clay? [1½+½+½+2+½]
 - b) How is porcelain manufactured? What are the methods used in the manufacture of porcelain? Write down the steps involved in the manufacture of porcelain by wet process. [1+1½+2½]
 - c) What are refractories? Explain the classification of refractories with one example of each type. Name two refractories whose refractoriness is about 2500°C. [1+3+1]
 - d) What is polysilazane? Give the method of manufacture of 'polysilazane' which is used as precursor to advanced ceramics (TDC). Mention its properties and uses. [1+3+1]
 - e) What is silica brick? Explain the different steps involved in the manufacture of silica bricks. Mention its properties and uses. Give a block diagram for the manufacture of refractories. [1+2+1+1]
 - f) Explain the following terms—
 - i) Thermal spalling
 - ii) Dimensional stability
 - iii) Pyrometric cone test
 - iv) Porosity
 - v) High-duty refractory brick [1+1+1+1+1]

3. Answer **any one** : [1×5]
 - a) What is the composition of ordinary glass? Briefly narrate the operation of a pot furnace in making optical glass. What is annealing process during shaping of glass articles? [1+3+1]
 - b) Name three important raw materials of glass manufacture with their chemical symbols. Distinguish between FORCAULT and COLBURN Process in shaping of sheet. Write composition of window glass. [1½+2½+1]
 - c) What is material science? What types of materials are produced by applying this technology? Give brief description about these materials. [5]

Group – B

(Answer **any two** questions)

[2× 5]

4. a) Equal weights of pure glucose ($C_6H_{12}O_6$) and acetaldehyde (C_6H_4O) are burnt in a combustion boat with pre oxygen to ensure complete combustion
Calculate and show which of the two compounds has higher calorific value (Heat of combustion of C and H are 8137 and 34500 Cal/g respectively)
- b) Imported Australian coal was analysed at custom laboratory with following results:
- | | |
|-----------------|------|
| | % |
| Moisture | 3.5 |
| Volatile matter | 30.0 |
| Ash | 12.0 |
- Calculate F.C% on drybasis, dry Ash free (daf) basis, and dry mineral matter free (dmmf) basis.[2½×2]
5. a) Write short notes on : “Peat” and “Brown coal” mentioning place of occurrence calorific value, and industrial application.
- b) Write how proximate analysis of coal is carried out in Laboratory. [2½+2½]
6. a) Enumerate the important fractions of HTC tar distillate with composition and boiling range.
What is coal tar pitch? State its use.
- b) What is metallurgical coke? Briefly describe its production process in coke oven plant. [2½+2½]

Group – C

Answer **any one** :

[1×10]

7. Write short notes on : Calcination , Roasting, Smelting, Gibbs Phase Rule [4×2½]
8. Draw Fe – Fe_3C phase diagram. Show all the phases in it and explain briefly. Show eutectic and eutectoid. [8+2]

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